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Fig. (4.8) Min. and Max. of radon concentrations in water samples.

4.3. Conclusions

The discussion of the results, which are obtained from this study leads to the following conclusions :

- 1- The uranium concentration varies from 1.57 to 2.73 ppm and from 0.67 to 2.12 ppm in soil and water samples respectively and less than the allowed limit (11.7 ppm).
- 2- The radon gas concentration varies from 226.89 to 665.56 Bq/m^3 and from 8.64 to 14.20 Bq/m^3 in soil and water samples respectively and less than the limit recommended (100 Bq/m^3).
- 3- The maximum uranium concentration in soils was (2.73 ppm) and in water(2.12ppm) in the Al-Taji and Al-Ashaar regions respectively .
- 4- The maximum radon gas concentration in soils was (665.56 Bq/m^3) and in water (14.20 Bq/m^3) in the Al-Taji and Al-Ashaar regions respectively.
- 5- Radionuclides (²²²Rn and ²³⁸U) had high concentration in site (S₄) and in site (W₉) because these sites are a field of a military operations during the war against Iraq in 2003.

4.4. Suggestions and Future Orientations

- 1- Study the radioactivity in soil and water by using γ -ray spectroscopy technique.
- 2- Study the uranium concentrations in biological samples for people living at these regions.
- 3- Study the uranium and radon gas concentrations for some weapons materials were using in military operations.

جمادي الأولى حزيران

Examination Committee Certificate

We certify that we have read the thesis entitled "Determination of Radioactivity in Soil and Water in Baghdad, Karbala and Basrah Samples" and as Examining Committee, examined the student Mustafa Arab Badai Al-Baidhani in its contents and what is related to it, and that in our opinion it is adequate as standard of thesis, with Excellent standing of degree of Master of Science in Physics.

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List of Symbols

Symbol	Description			
Am-Be	Americium –Beryllium Source			
Bq	Bequerel			
°C	Degree Centegrate			
Cs	uranium concentration of standard sample			
C _X	uranium concentration of unknown sample			
D	diameter of the etch pit			
DU	Depleted uranium			
E _B	Activation Energy of the Bulk Etching			
E _T	Activation Energy of the Track Etch			
EPA	Environmental Protection Agency			
h	thickness of the surface removed by etching			
IAEA	International Atomic Energy Agency			
ICRP	International Committee of Radiation Protection			
k	Boltzman constant = 1.38×10^{-23} j mol /K			
КОН	Potassium Hydroxide			
L	length of the latent track			
LSA	Low Specific Activity			
n	neutron			
N	Normality			
NaOH	Sodium Hydroxide			
ppm	Part Per million			

R	Length of track at the removed surface			
R.O.	Reverse Osmosis			
S	Soil sample			
SSNTDs	Solid State nuclear track detectors			
t	etching time			
Т	Temperature of the etching solution			
UNEP	United Nation Environment Program			
UNSCEAR	United Nation Scientific Committee on the Effects of Atomic			
	Radiation			
V	volume of distilled water			
V _B	bulk etch rate			
V _T	track etch rate			
W	the weight of NaOH			
W	Water sample			
W _{eq}	equivalent weight of NaOH			
WHO	World Health Organization			
Z	Atomic Number			

Symbol	Description			
α	Alpha particle			
Ø	Incident angle			
Ø _C	Critical angle			
η	Etching efficiency			
ρ	Track Density			
ρ_{S}	track density of standard sample			
ρ_X	track density of unknown sample			
μm	micrometer			
μg	microgram			

3.1. Introduction

This chapter describes the methods for estimation of depleted uranium and radon gas concentrations for the different samples of soil and water (natural and standards) from different sites in Iraqi regions .The materials and the apparatus which used included.

3.2. Collection of Samples

Two types of samples (soils and water) were collected from different sites in Baghdad city and other regions of Iraq ,as shown in Fig. (3.1),(3.2),(3.3) and (3.4).

3.2.1. Soil Samples

Samples of soils were collected from different sites in Iraq regions (bombardment regions during the war agianst Iraq in 2003) from depth (5 cm), as shown in Table (3.1) and Fig. (3.1), (3.3) and (3.4) from Baghdad, Karbala and Basrah cities respectively. Then the samples dried and cleaned from the doping grinds by using special sieve (0.27 mm in diameter).

3.2.2. Water Samples

Samples of water were collected from different sites in Iraqi regions . One litter volume of water was collected from some projects of water refinement on the rivers (Tigris and Shatt Al –Arab), also collected from different houses (drinking water), as shown in Table (3.2) and Fig. (3.2) and (3.4) from Baghdad and Basrah cities respectively .

Date of samples	Number of samples	Symbol	City	Location
22-11-2004	1	S_1	Baghdad	Abu Ghreab
22-11-2004	2	\mathbf{S}_2	Baghdad	Al Binuq
22-11-2004	3	S ₃	Baghdad	Al Binuq (Civilian Advocate ship)
25-11-2004	4	S_4	Baghdad	Al Taji
28-11-2004	5	S_5	Baghdad	Al Mansur
28-11-2004	6	S ₆	Baghdad	Al Mansur (Al Mamoon communications)
28-11-2004	7	S_7	Baghdad	Al Washash
15-12-2004	8	S_8	Baghdad	Al Sader City (Destroyed Houses)
5-1-2005	9	S ₉	Baghdad	Al Doura (Electrical Station)
14-1-2005	10	S ₁₀	Baghdad	Al Meshahda (Al Ishaqi Company)
20-2-2005	11	\mathbf{S}_{11}	Karbala	Al Abbas Street
20-2-2005	12	S ₁₂	Karbala	Al Abbas Street
20-2-2005	13	S ₁₃	Karbala	Industrial Region
20-2-2005	14	S ₁₄	Karbala	Al Quzwinia
12-3-2005	15	S ₁₅	Basrah	Al Twesa
13-3-2005	16	S ₁₆	Basrah	Al Ashaar
13-3-2005	17	S ₁₇	Basrah	Al Tememeyh
13-3-2005	18	S ₁₈	Basrah	Al Muafaqia (Residentieal Region)

Table (3 –1) Locations of soils samples in Baghdad city and south regions.

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Table (3–2) Locations of water samples in Baghdad city (Tigris river) city and south regions (Shatt Al-Arab) .

Date of samples	Number of samples	Symbol	City	Location
4-1-2005	1	\mathbf{W}_1	Baghdad	Al Adhmya (Al Adhmya Bridge)
5-1-2005	2	W ₂	Baghdad	Al Doura (Electrical Station)
5-1-2005	3	W ₃	Baghdad	Al Hasanian Bridge
4-1-2005	4	W_4	Baghdad	Maqam Al Kudur region (Bab Al muadham Bridge)
4-1-2005	5	W ₅	Baghdad	Al Sinaq Bridge
4-1-2005	6	W_6	Baghdad	Al Kadhmia (Al Aemaa Bridge)
13-3-2005	7	\mathbf{W}_7	Basrah	Al Baradaia (Shatt Al Arab)
13-3-2005	8	W_8	Basrah	Al Tanumah (Shatt Al Arab)
13-3-2005	9	W9	Basrah	Al Ashaar (Shatt Al Arab) (near the Ports)
14-3-2005	10	W_{10}	Basrah	Al Muafaqia (Drinking Water of Houses)
14-3-2005	11	W ₁₁	Basrah	Petrochemical Factory (Drinking Water of R.O.)

3.3. Materials and Apparatus 3.3.1. The Track Detectors

Commercially available sheets of CR-39 plastic which are presently known to be the most sensitive SSNTD and also characterized by low background were used in the present work.

These detector sheets of 250 μ m thick where cut into small piece each of 1cm × 1cm area. The present sheets of CR-39 were made by Pershore Moulding LTD Company, U.K. The detector sheets were stored at normal laboratory conditions.

3.3.2. The Etchant Solution

Sodium hydroxide solution with 6.25 normality has been used for the etching process. The preparation of normality is using [86]:

$$\mathbf{w} = \mathbf{w}\mathbf{e}\mathbf{q} \times \mathbf{N} \times \mathbf{V} \qquad \dots (3.1)$$

where:

w = the weight of NaOH needed to prepare the given normality.

 w_{eq} = equivalent weight of NaOH = addition of the atomic weight of Na, O and H = 40.

N = normality = 6.25.

V = volume of distilled water = 250 ml.

The etchant compartment has a volume of about 250 ml contains the NaOH solution with 6.25 N. This apparatus is closed assembly, except for small vent at the top of the condenser tube, which prevents any change of etchant normality (concentration) during the experiment due to evaporation. The etching was performed at 60° C while the etching time was 6 hours.

3.3.3.The Water Bath

An etching bath of the type "Labsco" (Germany) was used in the present work. It includes a thermostat, which can be operated over a range of 20^{0} C to 110^{0} C. However, in this study the chemical etching was carried out at 60^{0} C as, suggested previously [21]. The accuracy of regulation of temperature is better than $\pm 0.1^{0}$ C.

3.3.4.The Optical Microscope

The counting of all the chemically etched tracks was carried out using an optical microscope. It is a standard transmission type binocular research microscope (Bausch & Lomb, Japan). It is capable of giving magnifications of up to 400 x.

3.3.5. The Irradiation Source

For the irradiation test, an (Am–Be) source with flux 5×10^3 n/cm².s was used. It emits fast neutrons from the (α , n) reaction such as:

⁹Be +
$$\alpha$$
 \longrightarrow ¹²C + $_{0}^{1}n$ + 5.76 MeV ...(3.2)

This source consists of a rod of (Am–Be) surrounded by a paraffin wax. This wax is usually used for moderating the fast neutrons to thermal energies [87].

3.4. Experimental Details for Uranium Concentration Measurement

In the present work, the methods for measurement of uranium concentrations of the samples (soil and water), which is called the irradiation methods are :

3.4.1. Experimental Procedure for Soils Samples

Soil samples were taken from locations in Iraqi regions, as shown in Table (3-1) and Fig. (3.1),(3.3) and (3.4). Soil samples were dried and cleaned from the doping grinds using special sieve (0.27 mm in diameter).

0.5g of soil samples were mixed with 0.1g of methylcellulose powder ($C_6H_{10}O_5$)_n used as a binding material. The mixture was pressed into a pellet of 1cm diameter and 1.5mm thickness.

The pellets were covered with (CR-39) detector and put in a plate of paraffin wax at a distance of (5 cm) from the neutron source (Am-Be) as shown Fig. (3.5), with flounce of thermal neutron (3.024×10^9 n.cm⁻²) and flux (5 ×10³ n cm⁻² s⁻¹), to obtain induced fission fragments from the equation (3.3) [88] :

 $^{235}\text{U} + _{0}^{1}\text{n} \text{ thermal} \longrightarrow ^{236}\text{U} + \text{fission fragments } \dots (3.3)$

After the irradiation time (7 d), (CR-39) detectors were etched in (6.25N) NaOH solution at temperature of 60 0 C for (6 h), then the induced fission tracks density were recorded using the optical microscope.

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Fig. (3.5) The irradiation of the detectors and samples to the neutron source [89].

The density of the fission tracks (ρ) in the samples was calculated according to the following relation [90].

Track density (
$$\rho$$
) = Average number of total pits (tracks)
Area of field view ...(3.4)

The uranium concentrations in the soil samples were measured by comparison between track densities registed on the detectors of the sample pellet and that of the standard geological sample pellets from the relation [6,7,91] :

$$C_X$$
 (sample) / ρ_X (sample) = C_S (standard) / ρ_S (standard) ...(3.5)

$$C_{\rm X} = C_{\rm S} \, . \, (\rho_{\rm X} \, / \, \rho_{\rm S}) \, ...(3.6)$$

Fig. (3.6) shows this relation , when (slope = $\rho_{s}\,/\,C_{s}\,$) .

Where :

•

C_X : uranium concentration in unknown sample (ppm).

 C_S : uranium concentration in standard sample (ppm).

 ρ_X : track density of unknown sample (tracks/mm²).

 ρ_s : track density of standard sample (tracks/mm²).

Uranium concentration for standard samples (ppm)

Fig.(3.6) The relation between track density and uranium concentration (ppm) for standard geological soil samples.

3.4.2. Experimental Procedure for Water Samples

Water samples were taken from locations in Iraqi regions, as shown in Table (3-2) and Fig. (3.2) and (3.4).

Measurement of alpha-emitters in the environmental surface water can be complicated by the presence of a mass of suspended uranium bearing particle [13]. The procedure is to place known volume of water in the form of few drops (2 drops = 0.04 cc) on a track detector materials , water droplets are allowed to evaporate at room temperature , as sketched in Fig. (3.7), this leaves a thin residue of non-volatile constituents, the detector was then covered with a second piece of the detector. Subsequently exposure to (Am-Be) neutron source with flux of thermal neutrons (5×10^3 n/cm².s) for (7 d). Fig. (3.5) presents the irradiation of the detectors and the samples to the neutron source.

The detectors base or cover were etched in (6.25N) NaOH solution at temperature of 60 0 C for (6 h) and washed by disttle water, dried and the number of fission tracks produced was then counted by using the optical microscope.

This procedure was done to all samples including the standard samples

Fig. (3.7) Evaporation of the water droplet and the formation of a thin deposit.

The uranium concentrations in the water samples was measured by comparison between track densities of the water samples and that of the standard solutions from the equation (3.6).

The concentration – track density relationship for the present standard samples are illustrated in Fig.(3.8).

Uranium concentration for standard samples (ppm)

Fig.(3.8) The relation between track density and uranium concentration (ppm) for standard water samples.

3.5. Experimental Details for Radon Concentration Measurements

The radon concentrations in two types of samples(soil and water), were measured using the natural exposure method .

3.5.1. Experimental Procedure for Soils Samples

Soil samples were taken from locations in Iraqi regions, shown in Table (3-1) and Fig. (3.1),(3.3) and (3.4).

Soil samples were dried and cleaned from the doping grinds by using special sieve (0.27 mm in diameter). 10 g of soil samples were placed in plastic can. The dimensions of the can minimize the effect of gas thoron. Pieces of (CR-39) track detectors were fixed under the cover of plastic can , which contain the soil samples. The exposure time was (30 d) as shown in Fig. (3.9).

After the exposure time, (CR-39) detectors were etched in 6.25 N (NaOH) solution at temperature of 60 0 C for (6 h) and washed by disttel water, dried. The tracks density were recorded using the optical microscope with magnification (400 x).

Fig.(3.9)The apparatus of Radon gas (^{222}Rn) estimation by using (CR-39) detector for soil sample .

At the same way in determination of the uranium concentrations , radon gas (222 Rn) concentrations in the soil samples was measured from the equation (3.6) by comparison between tracks densities registed on the detectors around the sample and that of the standard geological sample .

Fig. (3.10) shows the relation between the tracks density and radon concentration for the standard samples .

Radon concentration for standard samples (Bq/m^3)

Fig.(3.10) The relation between track density and radon concentration (Bq/m³) for standard geological soil samples.

3.5.2. Experimental Procedure for Water Samples

Water samples were taken from locations in Iraqi regions , as shown in Table (3-2) and Fig. (3.2) and (3.4).

The method includes exposing the detectors to the water samples directly for certain period of time.

(250 ml) volume of water samples were placed in plastic can at room temperature . Pieces of (CR-39) track detectors with dimensions (1 ×1 cm²) were fixed under the cover of plastic can. The exposure time were (30 days) and the detection geometry was almost 4π , and the experimental set up is shown in Fig. (3.11).

After the exposure period, all the track detectors were etched in (NaOH)solution with 6.25 N at 60 0 C for (6 h). The detectors were then washed in distilled water, dried and were scanned under the optical microscope to record the tracks density .

Fig.(3.11) Radon gas (²²²Rn) estimation by using (CR-39) detector for water sample .

Radon gas (222 Rn) concentrations were determined by comparing mean track densities recorded in CR-39 detectors of water samples and with the standard samples from the equation (3.6).

Fig. (3.12) shows the relation between the track density and radon concentration (Bq/m^3) for standard sample .

Fig.(3.12) The relation between track density and radon concentration (Bq/m³) for standard water samples.

Examination Committee Certificate

We certify that we have read the thesis entitled "Determination of Radioactivity in Soil and Water in Baghdad, Karbala and Basrah Samples" and as Examining Committee, examined the student Mustafa Arab Badai Al-Baidhani in its contents and what is related to it, and that in our opinion it is adequate as standard of thesis, with Excellent standing of degree of Master of Science in Physics.

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Determination of the Radioactivity in Soil and Water in Baghdad, Karbala and Basrah Samples

A Thesis Submitted to the College of Science Al-Nahrain University in partial fulfillment of the requirements for the Degree of Master of Science in Physics

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